

a method similar to that used to prepare benzyl S-aziridine-2-carboxylate [4], was added to a solution of 0.222 g (10 mmole) of acrylate IX, and the mixture was heated at 60°C for 8 h. The solution was then evaporated, and the residue was separated by preparative liquid chromatography.

LITERATURE CITED

1. O. G. Nabiev, M. A. Shakhgel'diev, I. I. Chervin, and R. G. Kostyanovskii, *Dokl. Akad. Nauk SSSR*, **284**, 872 (1985).
2. A. V. Eremeev, F. D. Polyak, A. F. Mishnev, Ya. Ya. Bleidelis, É. É. Liepin'sh, Sh. S. Nasibov, I. I. Chervin, and R. G. Kostyanovskii, *Khim. Geterotsikl. Soedin.*, No. 11, 1495 (1982).
3. J. P. Jennings, W. Klyne, and P. M. Scopes, *J. Chem. Soc.*, No. 12, 7211 (1965).
4. K. Nakajima, F. Takai, T. Tanaka, and K. Okawa, *Bull. Chem. Soc. Jpn.*, **51**, 1577 (1978).

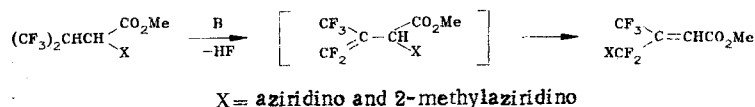
ADDUCTS OF METHYL β,β -BIS(TRIFLUOROMETHYL)ACRYLATE WITH NUCLEOPHILES AND ALLYLIC REARRANGEMENT DURING THEIR DEHYDROFLUORINATION

I. V. Solodin, A. V. Eremeev,
I. I. Chervin, and R. G. Kostyanovskii

UDC 547.717'722.3

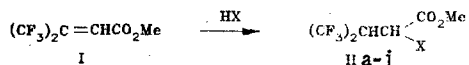
A number of adducts of methyl β,β -bis(trifluoromethyl)acrylate with various nucleophiles were synthesized. The boundaries of the allylic rearrangement of 2-substituted 3-trifluoromethyl-4,4-difluoro-3-butenates, obtained by dehydrofluorination of the cited adducts, were investigated.

We have previously observed allylic rearrangement with migration of an aziridinyll substituent [1]:



In the present research we investigated the boundaries of this rearrangement as a function of the types of substituents in the 2 position of the aziridinyll residue, as well as other heteroatomic α -substituents X (X = R¹R²N, MeO). For this, we synthesized adducts of β,β -bis(trifluoromethyl)acrylate I with a number of nucleophiles (Tables 1 and 2).

In addition to IIa, allylic rearrangement product III (12%) was isolated from the reaction mixture by high-performance liquid chromatography (HPLC) in the preparation of adducts IIa by heating an equimolar mixture of the starting reagents (50°C, MeCN), i.e., under the reaction conditions the dehydrofluorination of adduct IIa proceeds under the influence of nucleophile HX. Adducts IIb-i were obtained in 55-85% yields by the addition of the corresponding amines with cooling (-30°C, Et₂O). We were able to realize the addition of methanol only by heating (150°C) ester I in a sealed ampul with excess reagent in the presence of KF applied to Al₂O₃: adduct IIj was obtained in 55% yield.



II a X = 2-methoxycarboxylaziridino; b X = 2,2-dimethylaziridino; X = Me₂N; d X = MeO;
e X = azetidino; e X = pyrrolidino; g X = piperidino; h X = HOHN; i X = H₂N; j
X = Me₂NHN

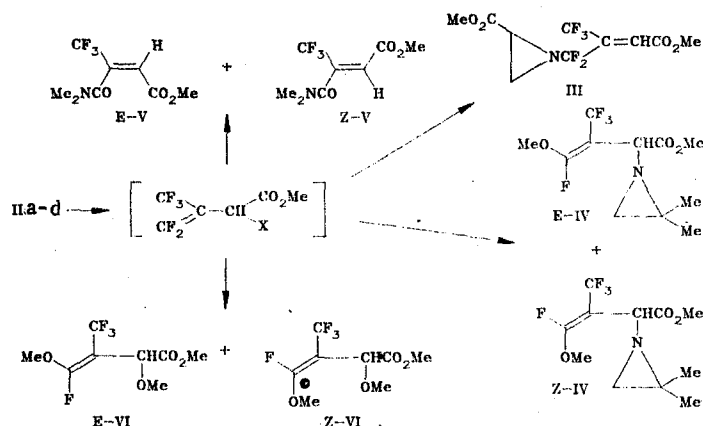
Institute of Organic Synthesis, Academy of Sciences of the Latvian SSR, Riga 226006.
Institute of Chemical Physics, Academy of Sciences of the USSR, Moscow 117977. Translated
from *Khimiya Geterotsiklicheskich Soedinenii*, No. 10, pp. 1359-1362, October, 1985. Original
article submitted January 23, 1985.

TABLE 1. Characteristics of the Synthesized Compounds

Compound	bp (mm) or mp, °C ^a	$\nu_{C=O}$, cm ⁻¹	Calc., %			Empirical formula	Found, %			Yield, %
			C	H	N		C	H	N	
IIa	50 (0,5)	1745, 1762	37,1	3,4	4,4	C ₁₀ H ₁₁ F ₆ NO ₄	37,2	3,5	4,5	57
IIb	55 (5)	1750	40,9	4,4	4,8	C ₁₀ H ₁₃ F ₆ NO ₂	40,5	4,5	4,9	79
IIc	45 (5)	1745	35,9	4,1	5,2	C ₈ H ₁₁ F ₆ NO ₂	35,7	4,2	5,3	65
IId	78 (40)	1755	33,1	3,1	—	C ₇ H ₈ F ₆ O ₃	33,3	3,2	—	55
IIe	50 (5)	1761	38,7	3,9	5,0	C ₉ H ₁₁ F ₆ NO ₂	38,9	4,0	4,9	82
IIf	55	1765	40,9	4,4	4,8	C ₁₀ H ₁₃ F ₆ NO ₂	40,7	4,3	4,9	71
IIf · HCl	94	1766	36,4	4,2	4,2	C ₁₀ H ₁₄ ClF ₆ NO ₂	36,6	4,1	4,4	95
IIg	62	1768	43,0	4,9	4,6	C ₁₁ H ₁₅ F ₆ NO ₂	43,1	4,8	4,5	68
IIg · HCl	95	1760	38,4	4,7	4,1	C ₁₁ H ₁₆ ClF ₆ NO ₂	38,6	4,6	4,2	95
IIh	34	1750	28,2	2,7	5,5	C ₈ H ₇ F ₆ NO ₂	28,1	2,5	5,4	85
IIi	59 (12)	1758	30,1	2,9	5,9	C ₈ H ₇ F ₆ NO ₂	30,2	2,8	5,8	76
IIj	70 (12)	1740	34,0	4,3	9,9	C ₈ H ₁₂ F ₆ N ₂ O ₂	34,3	4,4	9,8	79
III	60 (0,5)	1750, 1760	39,6	3,3	4,6	C ₁₀ H ₁₀ F ₅ NO ₄	39,9	3,2	4,7	12
E-IV		1755					43,4	5,3	5,2	21
Z-IV		1755	43,9	5,5	5,1	C ₁₁ H ₁₅ F ₄ NO ₃	43,7	5,4	5,3	12
E-V		1740					42,4	4,2	6,3	33
Z-V		1740	42,7	4,4	6,2	C ₈ H ₁₀ F ₃ NO ₃	42,6	4,3	6,1	11
E-VI		1712					39,4	4,0	—	26
Z-VI		1715	39,0	4,1	—	C ₈ H ₁₀ F ₄ O ₄	39,1	3,9	—	13

^aThe melting or boiling points for the E and Z isomers of IV-VI were not determined because of the small amounts (50 mg) of the substances.

The expected allylic migration is observed only in the case of IIc when adducts IIb-d are treated with ground potassium hydroxide in refluxing toluene; the rearrangement is accompanied by hydrolysis of the difluoromethylene group to a carbonyl group to give isomeric amides E- and Z-V:



Only vinyl substitution of the fluorine atom by a methoxy group in the intermediate olefin under the influence of the methanol formed in the hydrolysis of the ester group occurs in the case of adduct IIb to give isomeric esters E- (21%) and Z-IV (12%). In this case allylic migration of the 2,2-dimethylaziridinyl substituent is evidently hindered, and a competitive reaction with the less bulky nucleophile is therefore realized.

Only vinyl substitution of the fluorine atom by a methoxy group to give isomeric esters E- and Z-VI is also observed in the dehydrofluorination of adduct IId.

EXPERIMENTAL

The PMR spectra of 10% solutions of the substances in CDCl₃ were recorded with Bruker WP-80-SY (80 MHz), WH-90 (90 MHz), and WM-400 (400 MHz) spectrometers with tetramethylsilane (TMS) as the internal standard. The ¹⁹F NMR spectra were recorded with Perkin-Elmer R-12 (56 MHz) and Bruker WP-80-SY (75.39 MHz) spectrometers with CF₃COOH as the external standard. The IR spectra were obtained with a Perkin-Elmer 580-B spectrometer. Analytical and preparative liquid chromatography was carried out with a Du Pont Prep LC chromatograph with a UV spectrophotometer as the detector and Zorbax SIL 4.6 by 250 mm and Zorbax SIL 22.7 by 250 mm columns.

The characteristics of the synthesized IIa-j, III, and the E and Z isomers of IV-VI are presented in Table 1.

TABLE 2. Parameters of the PMR and ^{19}F PMR Spectra of II-VI

Com- pound	PMR spectrum ^a				^{19}F NMR spectrum		
	δ , ppm				J, Hz	δ , ppm	J, Hz
	COOMe	N-CH-CO	CH(CF ₃) ₂	other groups			
IIa	3,73; 3,82	2,87-3,20	3,91	1,66-2,18; 2,30-2,69 (CH-CH ₂)	8,0 ($^3J_{\text{HF}}$)	14,91; 17,07 (2CF ₃)	9,0 ($^4J_{\text{FF}}$); 9,0 ($^3J_{\text{FH}}$)
IIb	3,30	3,25	3,82	1,00 (2Me); 0,77-1,67 (CH ₂)	9,0 ($^3J_{\text{HF}}$); 2,7 ($^3J_{\text{HH}}$)	(A) 14,50; 18,40; (B) 15,70; 17,80	10,5 ($^4J_{\text{FF}}$); 9 ($^3J_{\text{FH}}$); 10,5 ($^4J_{\text{FF}}$); 9 ($^3J_{\text{FH}}$)
IIc	3,79	—	—	3,70 (ABX ₃); 2,31 (Me ₂ N)	11,0 ($^3J_{\text{HH}}$); 7,3 ($^3J_{\text{HF}}$) $J_{\text{AB}}=26,3$ 7,0 ($^3J_{\text{HF}}$)	13,66-14,21; 14,71-15,53 (3CF ₃)	8,54 ($^4J_{\text{FF}}$); 8,6 ($^3J_{\text{FH}}$)
II d	3,76	—	3,62	3,49 (MeO); 0,09 (HCO)	10,5 ($^3J_{\text{HH}}$); 7,8 ($^3J_{\text{HF}}$) $J_{\text{AB}}=97,5$	12,82; 16,39 (2CF ₃) 13,80-14,30; 15,40-15,90 (2CF ₃)	—
IIe	3,84	—	—	3,54-3,8 (CH-CH); 2,03; 3,67 (azetidino group)	—	—	—
II f	3,767	—	—	3,88 (ABX ₃ , CF ₃ -CH-CH); 1,72; 2,77; 2,58 (pyrrolidino group)	—	—	—
II f · HCl	3,91	5,00	4,64	2,07-2,42; 3,22-3,67; 3,70- 3,82 (pyrrolidino group)	8,0 ($^3J_{\text{HF}}$)	14,33; 18,22 (2CF ₃)	9,0 ($^4J_{\text{FF}}$); 9,0 ($^3J_{\text{FH}}$)
II g	3,78	3,49-3,69	3,74-3,98	1,23-1,71; 2,10-2,45; 2,49- 2,87 (piperidino group)	7,0 ($^3J_{\text{HF}}$)	13,80-14,10; 15,10-15,60 (2CF ₃) 13,41; 18,14 (2CF ₃)	—
II g · HCl	3,89	4,98	4,27	1,56-1,93; 1,98-2,40; 3,16- 3,80 (piperidino group)	—	—	—
II h	3,78	4,13	3,69	5,47 (NH, HO)	8,0 ($^3J_{\text{HF}}$); 4,0 ($^3J_{\text{HH}}$)	—	—
II i	3,80	3,75	4,09	1,75 (NH ₂)	8,0 ($^3J_{\text{HF}}$); 6,0 ($^3J_{\text{HH}}$)	—	—
II j	3,71	3,02	3,58	2,38 (Me ₂ N); 4,00 (NH)	8,0 ($^3J_{\text{HF}}$); 6,0 ($^3J_{\text{HH}}$)	—	—
II l	3,78; 3,84	—	—	2,31; 2,45; 3,08 (CH ₂ -CH); 6,82 (CH)	1,5 ($^4J_{\text{HF}}$)	—	—
E-IV	3,76	3,59	—	1,22; 1,27 (Me ₂ C); 1,20-1,30; 1,58 (CH ₂); 3,89 (MeOCF)	1,1 ($^4J_{\text{MeOCF}}$)	(A) 21,09 (CF ₃); -2,54 (CF)	11,0 ($^4J_{\text{FF}}$)
Z-IV	3,77	3,59	—	1,22; 1,27 (Me ₂ C); 1,20-1,30; 1,56 (CH ₂); 3,89 (MeOCF)	1,1 ($^4J_{\text{MeOCF}}$)	(B) 21,82 (CF ₃); -1,38 (CF)	10,0 ($^4J_{\text{FF}}$)
E-V	3,79	—	—	2,97; 3,07 (Me ₂ N); 6,44 (CH)	—	(A) 21,79 (CF ₃); -0,85 (CF)	18,8 ($^4J_{\text{FF}}$)
Z-V	3,84	—	—	3,04 (Me ₂ N); 6,24 (CH)	—	(B) 22,76 (CF ₃); -1,53 (CF)	20,0 ($^4J_{\text{FF}}$)
E-VI	3,82	—	—	3,45; 3,91 (2MeO); 4,56 (CH)	—	11,78 (CF ₃)	1,7 ($^4J_{\text{HF}}$)
Z-VI	3,80	—	—	3,45; 3,91 (2MeO); 4,57 (CH)	—	16,91 (CF ₃) 20,37 (CF ₃); -1,02 (CF)	11,4 ($^4J_{\text{FF}}$)
						21,02 (CF ₃); 1,78 (CF)	18,4 ($^4J_{\text{FF}}$)

^aThe letters A and B denote a diastereomeric pair; B/A = 2.6 for IIb (spectrum recorded at -10°C in d₈-toluene), and B/A = 1.7 for E- and Z-IV. Complex multiplets are observed for IIc, e, f, g.

Methyl 2-(2-Methoxycarbonylaziridino)-3-trifluoromethyl-4,4,4-trifluorobutanoate (IIa)
and Methyl 3-Trifluoromethyl-4-(2-methoxycarbonylaziridino)-4,4-difluoro-2-butenoate (III).
A mixture of 0.44 g (2 mmole) of methyl β,β -bis(trifluoromethyl)acrylate and 0.2 g (2 mmole) of methyl aziridine-2-carboxylate in 50 ml of acetonitrile was heated at 50°C for 8 h, after which it was evaporated, and the residue was separated by HPLC.

Methyl 2-(2,2-Dimethylaziridino)-3-trifluoromethyl-4,4,4-trifluorobutanoate (IIb). A solution of 10 mmole of 2,2-dimethylaziridine in 10 ml of ether was added at -30°C to 2.2 g (10 mmole) of methyl β,β -bis(trifluoromethyl)acrylate in 50 ml of ether, after which the mixture was maintained at this temperature for 0.5 h. It was then evaporated, and the residue was distilled in vacuo.

Adducts IIc, e-h, j were similarly obtained.

Methyl 2-Pyrrolidino-3-trifluoromethyl-4,4,4-trifluorobutanoate Hydrochloride (IIf•HCl)
and Methyl 2-Piperidino-3-trifluoromethyl-4,4,4-trifluorobutanoate Hydrochloride (IIg•HCl).
These compounds were obtained by passing gaseous HCl into solution of the corresponding adducts in absolute ether.

Methyl 2-Methoxy-3-trifluoromethyl-4,4,4-trifluorobutanoate (IIId). A mixture of 22.2 g (100 mmole) of acrylate I, 30 g of MeOH, and 5 g of KF/Al₂O₃ in a sealed ampul was heated at 150°C for 10 h, after which the mixture was filtered, and the filtrate was distilled in vacuo.

Methyl 2-Amino-3-trifluoromethyl-4,4,4-trifluorobutanoate (IIi). This compound was obtained by the method in [2].

(Z)- and (E)-Methyl 2-(2,2-Dimethylaziridino)-3-trifluoromethyl-4-fluoro-4-methoxy-3-butanoates IV. A 2.8-g (50 mmole) sample of ground KOH was added to a solution of 2 mmole of adduct IIb in 20 ml of absolute toluene, and the mixture was refluxed for 5 min. It was then cooled and diluted with an equal volume of ether, and the diluted mixture was filtered through a 1-cm-thick layer of silica gel. The filtrate was evaporated, and the residue was separated by preparative liquid chromatography.

Compounds (Z)- and (E)-V and (Z)- and (E)-VI were similarly obtained.

LITERATURE CITED

1. I. V. Solodin, A. V. Ereemeev, É. É. Liepin'sh, and R. G. Kostyanovskii, *Khim. Geterotsikl. Soedin.*, No. 4, 505 (1985).
2. I. L. Knunyants and Yu. A. Cheburkov, *Izv. Akad. Nauk SSSR, Otd. Khim. Nauk*, No. 12, 2162 (1960).